CE 329 Fall 2015

Class 8 Worksheet

Check that the mechanism is valid (is there a linear combination of the steps that equals the overall reaction?)
List the stable species:
List the reactive intermediates:
Set the rate of the overall reaction equal to the forward rate of the rate-determining step
If the rate expression contains concentrations of reactive intermediates, write equilibrium expressions for all steps except the rate-determining step
Solve the equilibrium expressions to get expressions for the concentrations of the reactive intermediates

A First Course on Kinetics and Reaction Engineering

Substitute the expressions for the reactive intermediate concentrations into the rate expression and simplify.	I
Repeat for the other two cases.	
Instead of assuming a rate determining step, use the Bodenstein steady state approximation.	